

ACROPTILIN - A NEW SESQUITERPENE LACTONE FROM
ACROPTILON REPENS

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Two crystalline substances have been isolated from the leaves and flower heads of Acroptilon repens (L.) DC. A new sesquiterpene lactone - repin - has been reported previously [1].

We give the results of a study of a second substance having the composition $C_{21}H_{24}O_9$ [1] with mp 196-198° C (from ethanol), $[\alpha]_D^{20} +92.3^\circ$ (c 0.688; alcohol); it contains $2H_{1ab}$; on chromatography in a thin layer of neutral alumina (activity grade IV) in the petroleum ether-benzene-chloroform-methanol (5:4:2:1) system it had R_f 0.29, in the benzene-ethanol (9:1) system R_f 0.44, and in ethyl acetate R_f 0.74; revealing agent a 0.5% solution of potassium permanganate in 0.5% sulfuric acid.

The IR spectrum has absorption bands in the 3470 region (OH group) and the 1743 cm^{-1} region (C=O) and a weak band at 1665 cm^{-1} (C=C). In the UV spectrum above 209 $m\mu$ there are no maxima characteristic for conjugated systems.

When the substance was heated with an ethanolic solution of alkali, three equivalents of caustic potash was consumed. However, the saponification products did not separate out. In the cold, the substance dissolves slowly in 5% caustic potash solution. From the neutral fraction obtained on saponification was obtained a colorless crystalline substance (II) with the composition $C_{15}H_{20}O_6$, mp 184-186° C (decomp). The IR spectrum of (II) has a well-defined peak at 3550 cm^{-1} (OH group), a broad band at 3440-3280 cm^{-1} (adjacent OH groups), and bands at 1775 (γ -lactone) and 1645 cm^{-1} (C=C). The IR spectrum of its acetyl derivative (III) has absorption bands at 3510 cm^{-1} (OH group), 1780 (γ -lactone), 1740 and 1255 ($OCOCH_3$), and 1650 cm^{-1} (C=C).

On hydrogenation, I consumed 3 moles of hydrogen per mole, and when it was dehydrogenated over selenium at 240-260° C for 2.5 hr guaiazulene was obtained.

These results show that the second substance isolated from Acroptilon repens is also a new sesquiterpene lactone; we have called it acroptilin.

REFERENCE

1. R. I. Evstratova, R. Ya. Rzazade, and K. S. Rybalko, KhPS [Chemistry of Natural Compounds], 290, 1966.

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TURCOMANIC ACID FROM JUNIPERUS TURCOMANICA

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Juniperus turcomanica B. Fedtsch. was collected in November 1965 on Mount Dushak close to Kheirabad in the Geok-Tepe region, Turkmen SSR.

When an acetonetic extract of the needles (after the elimination of acetone) was shaken in ethereal solution with a 2% solution of sodium carbonate, a solid suspension formed at the boundary between the two layers. The solution was filtered and the residue was washed with ether. After recrystallization from ethanol, colorless crystals with mp 356-357° C were isolated. The crystals were dissolved in water and the aqueous solution was acidified with dilute sulfuric acid. A colorless precipitate readily soluble in diethyl ether deposited.